

Sweet Sixteen Chemistry Compound Tournament Answers



Sweet Sixteen Chemistry Compound Tournament

tournament. This clever activity combines the ever-popular March Madness basketball pool with a review of chemical formulas, the structure and charges of ions, and molar mass calculations. The result is the Sweet 16 Chemistry Compound Tournament. Hopefully, your students will enjoy playing the "Tournament" while reviewing these important topics.

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Sweet 16 Chemistry Ion Tournament First Round Second Round Semifinals Finals Winner NiCl₂ Ni(II)₂+ Fe(II) S Zn CO₃ Na CrO₄ Mg PO₄ Ag(I) SO₄ Ba NO₃ Al OH Rules 1. First Round →Add Charges 2. Second Round →Write Formula 3. Semis →Soluble Compound (compound that will dissolve in water) 4. Finals →Soluble Product of Chemical Reaction 5.

Name Class/Period Sweet 16 Chemistry Ion Tournament

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Potassium phosphate First Round Second Round Semifinals Finals Winner Rules 1. First Round — Write formula, most number of atoms in formula unit wins 2. Second Round — Greater number of

ions in the formula unit wins 3. Semis — Higher charge on anion wins 4. Finals — The larger molar mass wins Sweet 16 Chemistry Compound Tournament

Potassium phosphate First Round Second Round Semifinals ...

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SWEET SIXTEEN CHEMISTRY COMPOUND TOURNAMENT ANSWERS

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