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Chemistry - Wikipedia

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The MO theory also works well for larger molecules such as N_2 , O_2 , CO_2 , and BF_3 and for ions such as NO_3^- and CO_3^{2-} , in which the bonding MOs are delocalized, involving three or more atoms. It is especially useful for aromatic molecules such as benzene.

Molecular Orbital Theory - Chemistry Encyclopedia ...

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Thermodynamics - Chemistry Encyclopedia - reaction, gas ...

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Papers in Press - The Journal of Biological Chemistry

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Chirality - Wikipedia

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explosive. It can also be used to prepare diazonium salts, which combine with anilines and phenols to form brightly colored azo compounds. The latter type of reaction can be used to produce azo-dyes, as well as serving as a qualitative test for aromatic amines.

Nitrous acid - New World Encyclopedia

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Angewandte Chemie International Edition: Early View

Selected Recent Research Paper R Guillet-Nicolas, R Ahmad, KA Cychosz, F Kleitz, M Thommes, "Insights into the pore structure of KIT-6 and SBA-15 ordered mesoporous silica—recent advances by combining physical adsorption with mercury porosimetry", *New Journal of Chemistry* 40 (5), 4351-4360 (2016) F Velasco, R Guillet-Nicolas, G Dobos, M Thommes, P Lodewyckx, "Towards a better understanding ...

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Scientists alive today who accept the biblical account of ...

adj. 1846, "of or like the nucleus of a cell," from nucleus + -ar, probably by influence of French nucléaire. Use in atomic physics is from 1914; of weapons, from 1945. Hence nuclear physics (1933), nuclear energy (1941), nuclear war (1954). Nuclear winter coined by Richard Turco, but first attested in article by Carl Sagan in "Parade" magazine, Oct. 30, 1983.

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