

Ncert Chemistry 12 Intext Answer Chapter 2



Ncert Chemistry 12 Intext Answer

3.7 Define conductivity and molar conductivity for the solution of an electrolyte. Discuss their variation with concentration. Sol. Conductivity of a solution is defined as the conductance of a solution of 1 cm length and having 1 sq. cm as the area of cross-section. Molar conductivity of a solution at a dilution (V) is the conductance of all the ions produced from one mole of the electrolyte ...

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NCERT Solutions Class 12 Chemistry UNIT 1 THE SOLID STATE Intext Questions. NCERT Solutions, CBSE Sample paper, Latest Syllabus, NCERT Books, ... Question 1.12: Distinguish between (i) Hexagonal and monoclinic unit cells (ii) Face-centred and end-centred unit cells.

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Class 12 Chemistry. Solution In Text Questions 1. Question - 2.1 - Calculate the mass percentage of benzene (C_6H_6) and carbon tetrachloride (CCl_4) if 22 g of benzene is dissolved in 122 g of carbon tetrachloride.. Answer: Given, Mass of solute (W B) = 22g Mass of solvent (W B) = 122 g. Mass of solution = 22g + 122 g = 144 g

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